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# ■ The effect of warming climate on nutrient and solute export from the Greenland Ice Sheet

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## **Abstract**

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Glacial meltwater runoff is likely an important source of limiting nutrients for downstream primary producers. This has particular significance for regions surrounding the Greenland Ice Sheet, which discharges >400 km³ of meltwater annually. The Arctic is warming rapidly but the impact of higher discharge on nutrient export is unknown. We present four years of hydrological and geochemical data from a large Greenland Ice Sheet catchment that includes the two highest melt years on record (2010, 2012). Measurements reveal significant variation in dissolved solute (major ion) and estimated dissolved macronutrient (nitrogen, phosphorus and silica) fluxes, with increases in higher melt years. Labile particulate macronutrients dominate nutrient export, accounting for ~50 % of nitrogen and >80 % of both phosphorus and silica. The response of ice sheet nutrient export to enhanced melting is largely controlled by particle bound nutrients, the future supply of which is uncertain. We propose that the Greenland Ice Sheet provides an underappreciated and annually dynamic source of nutrients for the polar oceans, with changes in meltwater discharge likely to impact marine primary productivity in future decades.

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## Introduction

Recent estimates predict global mean surface warming of up to  $4.8\,^{\circ}\text{C}$  above the 1986-2005 mean by 2100, with the polar regions subject to more extreme increases (Collins *et al.*, 2013). Already, the Greenland Ice Sheet has experienced

increased surface temperatures, with the five highest melt seasons on record occurring since 2000 (Tedesco *et al.*, 2013). In 2012, surface melting was the most widespread in over 100 years (Tedesco *et al.*, 2013). By 2100, the annual freshwater flux from the Greenland Ice Sheet could exceed 1000 km<sup>3</sup> a<sup>-1</sup>, making it one of the world's largest sources of freshwater (Fettweis *et al.*, 2013).

Currently, we lack information about the impact of meltwater on down-stream biogeochemical cycles, even though the coastal waters surrounding the ice sheet harbour highly productive ecosystems, that are strong  $CO_2$  sinks (Rysgaard et al., 2012). Recent work has highlighted the importance of glacier meltwater, including delivery of essential nutrients to the polar oceans (Bhatia et al., 2013; Wadham et al., 2013; Hawkings et al., 2014; Lawson et al., 2014). However, whether glacier melting provides an important negative climate feedback through its effect on marine primary production and  $CO_2$  drawdown remains unknown.

Future changes to Greenland Ice Sheet hydrology will probably impact the export of solute and reactive sediments to the polar oceans. Much of the meltwater drains from the surface to the ice sheet bed, chemically weathering the subglacial sediments (Bartholomew *et al.*, 2011). Supraglacial lake drainage events are particularly important because they rapidly channel large quantities of meltwater to the ice sheet bed, flushing out stored, solute-rich, subglacial waters (Bartholomew *et al.*, 2011; Hawkings *et al.*, 2014). Supraglacial lake formation and the migration of drainage systems inland in a warming climate (Leeson *et al.*, 2015) could expose new subglacial areas to meltwater flushing, potentially enhancing solute evacuation. Glaciers are effective at fracturing and grinding bedrock (Cowton *et al.*, 2012), producing turbid meltwaters with abundant, very fine suspended particles, *i.e.* >1 g L<sup>-1</sup> (Cowton *et al.*, 2012). Suspended material has recently been identified as a potential source of labile nutrients to near coastal regions (Hodson *et al.*, 2004; Bhatia *et al.*, 2013; Hawkings *et al.*, 2014; Wehrmann *et al.*, 2014) but data are sparse.

In this study, we present a full suite of geochemical and hydrological data from Leverett Glacier, a large (~600 km²), land terminating, outlet glacier of the Greenland Ice Sheet (details in Supplementary Information). The data cover four years (2009-2012) where melting intensity varied (Fig. 1), including the two highest melt seasons on record. This is the most comprehensive dataset yet on major ion and nutrient concentrations from a glacial system.

## **Results and Discussion**

Hydrological data (discharge, electrical conductivity and suspended material concentration) were collected for all four years (2009-2012) at a stable bedrock section ~2.2 km downstream of the glacier mouth. Major ion data are available for 2009, 2010 and 2012 and we used them to interpolate concentrations for 2011. Nutrient flux for 2009, 2010 and 2011 was estimated from 2012 data using a correlation with electrical conductivity. Further details are provided below and in Supplementary Information.



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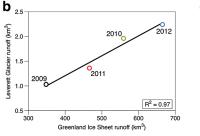
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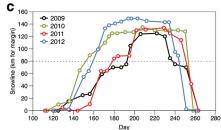
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97





40 km

Letter

Figure 1 Field site and basic hydrological data. (a) Leverett Glacier with the snowline transect used to determine the point where snow covers the ice surface, derived from MODIS imagery. Dashed lines represent elevation contours. The catchment area was determined from a surface digitial elevation model (Palmer et al., 2011). (b) Leverett Glacier meltwater discharge (Table 1) versus modelled Greenland Ice Sheet runoff (Tedesco et al., 2013). (c) Leverett Glacier snowline; open circles represent observed position; connecting lines are linear estimates of retreat, interpolated between the measurements. Estimated catchment extent is represented by the dotted line at ~80 km.

# Hydrology

The 2010 and 2012 ablation seasons produced the largest volumes of meltwater on record (Tedesco *et al.*, 2013). This is reflected in Leverett Glacier discharge (Fig. S-1), which was proportional to annual ice sheet runoff ( $R^2 = 0.97$ ; Fig. 1b). The snowline also reached maximum elevation in 2012, 14 km further inland than in 2010 and 2011 (Table 1; Fig. 1c). 2009 and 2011 can be considered "average" melt years, with discharge proportional to the mean meltwater flux over the past decade. 2010 and 2012, with significantly above average discharge, were "extreme" melt years. This characterisation provides a benchmark for evaluating future trends because the frequency of extreme seasons is likely to increase (Fettweis *et al.*, 2013).

Geochem. Persp. Let. (2015) 1, 94-104 doi: 10.7185/geochemlet.1510



Table 1 Flux and hydrological data.

	Units	Year			
		2009	2010	2011	2012
Greenland Ice Sheet runoff *	km³	348	559	466	665
Leverett Glacier discharge	km³	0.94	1.79	1.10	2.03
Snowline above catchment	est. days	45	100	73	85
Snowline retreat from margin	km	125	135	135	149
Solute Flux	eq	$3.0 \times 10^{8}$	$5.6 \times 10^{8}$	$3.2 \times 10^{8}$	$5.6 \times 10^{8}$
Sediment Flux	t	$3.7 \times 10^{6}$	$2.6 \times 10^{6}$	$3.0 \times 10^{6}$	$2.2 \times 10^{6}$
Dissolved inorganic nitrogen †	t	25	46	26	48
Dissolved silica †	t	130	230	120	230
Dissolved inorganic phosphorus†	t	6.2	12	6.7	12
Dissolved inorganic nitrogen ‡	t	22	41	25	47
Dissolved silica ‡	t	110	220	130	240
Dissolved inorganic phosphorus‡	t	7.8	15	8.9	16
Exchangeable NH <sub>4</sub> **	t	19-58	13-41	15-47	11-35
Amorphous silica**	t	18,000- 44,000	12,000- 31,000	14,000- 36,000	11,000- 26,000
NaOH extractable phosphorus**	t	20-130	14-92	16-110	12-78

All estimates are shown to Decimal Day 230/231, i.e. 17 August.

Estimates are reported with 2 significant digits.

eq = molar equivalent

t = tons of dry element

Snowline: the boundary where snow covers the underlying ice. Down glacier from this point is exposed ice, where the snow cover has melted.

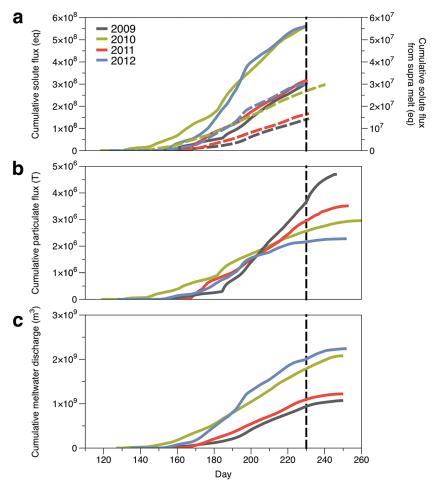
- \* Greenland Ice Sheet runoff estimates from Tedesco et al. (2013).
- \*\* Sediment fluxes given as range based on minimum and maximum extractable nutrient concentrations.
- † Fluxes estimated with electrical conductivity.
- ‡ Fluxes estimated using discharge weighted mean.

#### Solute flux

We estimated total solute export for all years from the electrical conductivity (EC) of the meltwater (Fig. S-3). The major ion ( $Ca^{2+}$ ,  $Mg^{2+}$ ,  $K^+$ ,  $Na^+$ ,  $Cl^-$ ,  $SO_4^{2-}$ ,  $HCO_3^-$ ) concentrations are a linear function of conductivity with  $R^2 = 0.87$  (2009), 0.77 (2010) and 0.98 (2012) (Figs. S-3 and S-4). We have a full hydrological dataset (discharge, EC and suspended material) from 2011 but major ion data are missing so using the concentration-conductivity correlation for data from 2009, 2010 and 2012 (n = 368;  $R^2 = 0.89$ ), we estimated 2011 solute concentrations (Fig. S-3). Solute flux was estimated by cumulatively summing the total solute concentration

99

at each conductivity measurement time step (Fig. 2). We differentiated supraglacial solute from the total solute flux to assess the importance of subglacial sources (Fig. 2a). The estimates suggest that flushing of stored subglacial waters and rapid weathering of subglacial sediments by dilute supraglacial meltwater account for >95 % of the solute export from the catchment.



**Figure 2** Cumulative solute and sediment flux. **(a)** Solid lines show solute flux (Ca<sup>2+</sup>, Mg<sup>2+</sup>, K<sup>+</sup>, Na<sup>+</sup>, Cl<sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, HCO<sub>3</sub>); dashed lines represent the estimated portion originating from supraglacial meltwater (right axis). **(b)** Cumulative particle borne flux. **(c)** Cumulative meltwater discharge. Day 230 (dashed line) was the limit of the dataset interpretation. Values at the dashed line correspond to data presented in Table 1 and Figure 1b.

A key discovery is that solute flux during the two extreme melt years was ~90 % higher than for the average years, indicating that solute export scales with discharge. Thus, it is likely that increased melting will increase solute fluxes from the Greenland Ice Sheet. Most of the solute is delivered during peak melt periods during the ablation season (Fig. S-5). Solute pulses are also released during supraglacial lake drainage events (Fig. S-5; Bartholomew *et al.*, 2011). Enhanced solute discharge during extreme years likely results from i) drainage of concentrated subglacial waters stored in poorly connected regions, *e.g.*, deeper into the ice sheet, ii) higher flushing rates and iii) rapid weathering of reactive subglacial sediments by large volumes of supraglacial meltwater. We next ascertain if nutrient fluxes follow the same trend.

#### Dissolved nutrient fluxes

Nutrient fluxes for all years were estimated from the 2012 dataset, the only complete macronutrient record available. For the most common, namely NH<sub>4</sub><sup>+</sup>, NO<sub>3</sub><sup>-</sup>, Si and PO<sub>4</sub><sup>3</sup>-, we also estimated annual flux by correlation with conductivity, as for major ion concentrations. This is justified by the good correlation of conductivity with nutrient concentrations (Fig. S-6). If the function holds for 2012, we assume it also holds for the other years. Si and P are released during rock weathering and dissolved inorganic nitrogen is also enhanced by subglacial biogeochemical processes, such as microbially mediated nitrification (NO<sub>3</sub><sup>-</sup>) and mineralisation of organic matter (NH<sub>4</sub>; Wynn et al., 2007). For example, mean nitrogen concentrations attributable to subglacial processes (~1 µM) are similar to supraglacial processes (~1.3 µM). Nitrogen correlates linearly with conductivity  $(R^2 = 0.74; Fig. S-6)$ . To account for the seasonal evolution in meltwater composition, we used two regressions (Fig. S-6) for dissolved silica ( $R^2 = 0.72$  for early season and  $R^2 = 0.34$  for bulk season runoff), and phosphate ( $R^2 = 0.61$  and 0.59, respectively). We propose that these differences arise from the change in source of the subglacial water as the melt season progresses, i.e. close to ice margins early in the season to more isolated inland subglacial waters as the season progresses. The source influences the subglacial flowpath length, hence water residence times and pH/redox conditions.

Higher dissolved nutrient flux correlates with higher discharge years. Inorganic nitrogen (86 %  $\pm$  9.8 %), dissolved silica (85 %  $\pm$  0.1 %) and phosphate (86 %  $\pm$  11.9 %) are higher in extreme melt years than for average years. This is significant and demonstrates the potential for nutrient release by a warming climate.

A different approach for estimating nutrient flux is to combine discharge weighted mean concentrations and total discharge flux (Hawkings  $et\ al.$ , 2014). Estimates from this method and the EC-based method are similar (Table 1). The weighted mean dissolved nitrogen and silica fluxes were marginally lower (~7 % and ~5 %) and phosphate flux was higher (~23 %), probably because of late melt season influence, when phosphate concentrations were high and where the bulk of the discharge occurred.



101

An important assumption is that the correlation of nutrient concentration with conductivity is consistent over the years. We have a limited dataset for  $NO_3^-$  and Si from 2009. Results are sparse so estimates are crude but they serve as a benchmark for comparison. Flux derived from 2009 data for  $NO_3^-N$  is 12 tons, compared with 15 tons estimated using 2012 data. Flux for dissolved Si is 180 tons, compared with 130 tons estimated using 2012 measurements (Table 1). Both are well within an order of magnitude, which offers confidence that our estimates from 2012 data are reasonable.

## Nutrient flux on particles

Glaciers effectively fracture and grind bedrock into high surface area, highly reactive, clay and silt sized particles, some of which are transported in runoff as suspended material (Gurnell and Clark, 1987; Brown *et al.*, 1996). By using data derived from the labile nutrients in the 2012 suspended material (n > 25), we estimated the range of nutrient concentrations associated with the particulate fraction (Table 1; Fig. 3). We assumed that the 2009-2011 mean extractable nutrient concentrations lie within the 2012 minimum and maximum concentrations, which is reasonable because the runoff comes from the same catchment and the mineral composition is relatively constant (Hawkings *et al.*, 2014).

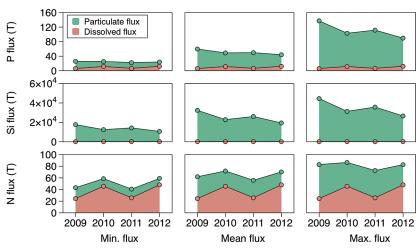


Figure 3 Estimated nutrient flux. Minimum (left column), mean (middle) and maximum (right) possible values give an impression of the range for phosphate, silicate and nitrogen compounds. Dissolved flux was determined using the electrical conductivity correlation.

Particulate bound nutrients account for a large portion of the estimated nutrient flux (Table 1; Fig. 3), which is significant in all years. The particulate transported fraction correlates with the nutrient source (lithogenic or atmospheric)

Geochem. Persp. Let. (2015) 1, 94-104 doi: 10.7185/geochemlet.1510

and the tendency for the ion to complex. Nutrients derived directly from rocks associate more with solids, *i.e.* for Si, >99 % and for P, ~80 % of the total flux. Nitrogen, which has a supraglacial component, is transported less on solids, *i.e.* <50 %. Our results are consistent with the low solubility of Si and the high affinity for P absorption onto solids, such as iron (oxyhyd)roxides. Ammonium is only weakly absorbed and nitrate remains preferentially in solution. This suggests that annual nitrogen flux is more sensitive to changes in ice sheet water discharge than particle flux.

We have demonstrated that, as in riverine systems (Mayer *et al.*, 1998; Ruttenberg, 2014), a high fraction of ice sheet nutrient export is associated with suspended material. This is consistent with previous research from Arctic glaciers (Hodson *et al.*, 2004) and supports recent assertions that the impact of terrigeneous material on the oceans is underestimated in global element cycling (Jeandel and Oelkers, 2015). Our results underline the need for more information about ice sheet sediment flux dynamics. As in previous studies, we observed highly variable annual sediment fluxes, which do not correlate well with discharge on a catchment basis (Fig. 2; Gurnell and Clark, 1987). Sediment flux might be less influenced by total meltwater discharge and more sensitive to meltwater access to fresh, subglacial sediment sources (Cowton *et al.*, 2012). However, evidence from past deglaciation events indicates that climate warming increases sediment export (Jeandel and Oelkers, 2015) and analysis of sediment plumes from meltwater rivers demonstrates a higher sediment flux in recent years (Hudson *et al.*, 2014).

The extent of biological consumption of the nutrients bound to particles before deposition and subsequent burial is unknown. Particulates from meltwater are extremely fine, *e.g.*, >95 % of particles can be <32 µm in size (Brown *et al.*, 1996), so surface area is high and nutrient transport in the buoyant, fresh water plumes in near coastal regions is likely to be significant. Evidence from recent polar studies shows that particle borne nutrients are carried far offshore (Schroth *et al.*, 2014; Wehrmann *et al.*, 2014) and nutrients deposited with glacial sediments in fjords can be resuspended in the water column (Wehrmann *et al.*, 2014). However, the scarcity of data means that the contribution of particle bound nutrients on oceanic productivity near Greenland remains uncertain.

Terrestrial and marine studies have shown that large fractions (75-95 %) of amorphous Si can be dissolved and recycled (Treguer *et al.*, 1995; Gibson *et al.*, 2000). Amorphous Si is an order of magnitude more soluble in saline solutions than in fresh water (Icenhower and Dove, 2000; Loucaides *et al.*, 2008) and recycling is favoured in estuaries (Loucaides *et al.*, 2008). NaOH extractable phosphate is commonly termed "algae available" (DePinto *et al.*, 1981) and its bioavailability has previously been demonstrated (Bostrom *et al.*, 1988). High salinity in ocean and fjord waters also favours P and NH<sub>4</sub> desorption (Garner *et al.*, 1991; Hodson *et al.*, 2004; Zhang and Huang, 2011), enhancing their bioavailability. Thus, annual sediment flux is an important factor in downstream productivity.



Letter

## Conclusions

Changes in the hydrological output from the Greenland Ice Sheet in a warming climate could have significant effect on solute and nutrient delivery to near coastal regions. Our data, from Leverett Glacier, a large representative ice sheet catchment, indicate that bulk solute and dissolved nutrient fluxes will increase as "extreme" melt year frequency increases. A significant fraction of nutrients, especially silica and phosphorous, will be transported by suspended particles. The extent of their influence depends on desorption before burial, bioavailability and change in the ice sheet particulate flux, which are currently uncertain. Our study demonstrates that retreating snowline and higher meltwater input into less efficiently drained subglacial regions are likely to increase the dissolved macronutrient flux. Particle bound nutrients have been largely overlooked but contribute significant mass to nutrient cycling. Increased warming, thus increased meltwater runoff, will likely impact regional nutrient availability, and thus, the carbon cycle.

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# **Additional Information**

**Supplementary Information** accompanies this letter at www.geochemicalperspectivesletters.org/article1510

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Geochem. Persp. Let. (2015) 1, 94-104 doi: 10.7185/geochemlet.1510

#### **Author Contributions**

All authors contributed. JLW and MT conceived the project. JRH, EL, AS, TC, IB, AT, PN, DC and JLW collected field data. JRH, JRH, EL and JT undertook the lab analysis. JRH, JLW and MT wrote the paper.

## References

- Bartholomew, I., Nienow, P., Sole, A., Mair, D., Cowton, T., Palmer, S., Wadham, J. (2011) Supraglacial forcing of subglacial drainage in the ablation zone of the Greenland ice sheet. Geophysical Research Letters 38, L08502.
- BHATIA, M.P., KUJAWINSKI, E.B., DAS, S.B., BREIER, C.F., HENDERSON, P.B., CHARETTE, M.A. (2013) Greenland meltwater as a significant and potentially bioavailable source of iron to the ocean. Nature Geoscience 6, 274-278.
- BOSTROM, B., PERSSON, G., BROBERG, B. (1988) Bioavailability of different phosphorus forms in freshwater systems. Hydrobiologia 170, 133-155.
- Brown, G.H., Tranter, M., Sharp, M.J. (1996) Experimental investigations of the weathering of suspended sediment by Alpine glacial meltwater. Hydrological Processes 10, 579-597.
- Collins, M., Knutti, R., Arblaster, J., Dufresne, J.-L., Fichefet, T., Friedlingstein, P., Gao, X., Gutowski, W.J., Johns, T., Krinner, G., Shongwe, M., Tebaldi, C., Weaver, A.J., Wehner, M. (2013) Long-term Climate Change: Projections, Commitments and Irreversibility. In: Stocker, T.F., Qin, D., Plattner, G.-K., Tignor, M., Allen, S.K., Boschung, J., Nauels, A., Xia, Y., Box, V., Midgley, P.M. (Eds.) Climate Change 2013: The Physical Science Basis. Contributions of Working Group I to the Fifth Assessment Report of the Intergovernmental Panel on Climate Change. Cambridge University Pess, Cambridge, UK and New York, NY, USA.
- COWTON, T., NIENOW, P., BARTHOLOMEW, I., SOLE, A., MAIR, D. (2012) Rapid erosion beneath the Greenland ice sheet. Geology 40, 343-346.
- DEPINTO, J.V., YOUNG, T.C., MARTIN, S.C. (1981) Algal-available phosphorus in suspended sediments from Lower Great Lakes Tributaries. Journal of Great Lakes Research 7, 311-325.
- Fettweis, X., Franco, B., Tedesco, M., van Angelen, J.H., Lenaerts, J.T.M., van den Broeke, M.R., GALLEE, H. (2013) Estimating the Greenland ice sheet surface mass balance contribution to future sea level rise using the regional atmospheric climate model MAR. Cryosphere 7, 469-489.
- GARDNER, W., SEITZINGER, S., MALCZYK, J. (1991) The effects of sea salts on the forms of nitrogen released from estuarine and freshwater sediments: Does ion pairing affect ammonium flux? Estuaries 14, 157-166.
- GIBSON, C.E., WANG, G., FOY, R.H. (2000) Silica and diatom growth in Lough Neagh: the importance of internal recycling. Freshwater Biology 45, 285-293.
- GURNELL, A.M., CLARK, M.J. (1987) Glacio-fluvial sediment transfer: An Alpine Perspective. Wiley & Sons, Chichester
- HAWKINGS, J.R., WADHAM, J.L., TRANTER, M., RAISWELL, R., BENNING, L.G., STATHAM, P.J., TEDSTONE, A., NIENOW, P., LEE, K., TELLING, J. (2014) Ice sheets as a significant source of highly reactive nanoparticulate iron to the oceans. Nature Communications 5.
- HODSON, A.J., MUMFORD, P., LISTER, D. (2004) Suspended sediment and phosphorus in proglacial rivers: bioavailability and potential impacts upon the P status of ice-marginal receiving waters. Hydrological Processes 18, 2409-2422.
- Hudson, B., Overeem, I., McGrath, D., Syvitski, J.P.M., Mikkelsen, A., Hasholt, B. (2014) MODIS observed increase in duration and spatial extent of sediment plumes in Greenland fjords. The Cryosphere 8, 1161-1176.



- ICENHOWER, J.P., DOVE, P.M. (2000) The dissolution kinetics of amorphous silica into sodium chloride solutions: Effects of temperature and ionic strength. Geochimica et Cosmochimica Acta 64, 4193-4203.
- Jeandel, C., Oelkers, E.H. (2015) The influence of terrigenous particulate material dissolution on ocean chemistry and global element cycles. *Chemical Geology* 395, 50-66.
- LAWSON, E.C., BHATIA, M.P., WADHAM, J.L., KUJAWINSKI, E.B. (2014) Continuous summer export of nitrogen-rich organic matter from the Greenland Ice Sheet inferred by ultra high resolution mass spectrometry. *Environmental Science & Technology* 48, 14248-14257.
- LEESON, A.A., SHEPHERD, A., BRIGGS, K., HOWAT, I., FETTWEIS, X., MORLIGHEM, M., RIGNOT, E. (2015) Supraglacial lakes on the Greenland ice sheet advance inland under warming climate. *Nature Climate Change* 5, 51-55.
- LOUCAIDES, S., VAN CAPPELLEN, P., BEHRENDS, T. (2008) Dissolution of biogenic silica from land to ocean: Role of salinity and pH. *Limnology and Oceanography* 53, 1614-1621.
- MAYER, L.M., KEIL, R.G., MACKO, S.A., JOYE, S.B., RUTTENBERG, K.C., ALLER, R.C. (1998) Importance of suspended particulates in riverine delivery of bioavailable nitrogen to coastal zones. Global Biogeochemical Cycles 12, 573-579.
- Palmer, S., Shepherd, A., Nienow, P., Joughin, I. (2011) Seasonal speedup of the Greenland Ice Sheet linked to routing of surface water. *Earth and Planetary Science Letters* 302, 423-428.
- RUTTENBERG, K. (2014) The global phosphorus cycle. Treatise on Geochemistry 8, 499-558.
- Rysgaard, S., Mortensen, J., Juul-Pedersen, T., Sørensen, L.L., Lennert, K., Søgaard, D.H., Arendt, K.E., Blicher, M.E., Sejr, M.K., Bendtsen, J. (2012) High air—sea CO2 uptake rates in nearshore and shelf areas of Southern Greenland: Temporal and spatial variability. *Marine Chemistry* 128–129, 26–33.
- SCHROTH, A.W., CRUSIUS, J., HOYER, I., CAMPBELL, R. (2014) Estuarine removal of glacial iron and implications for iron fluxes to the ocean. *Geophysical Research Letters* 41, 3951-3958.
- Tedesco, M., Fettweis, X., Mote, T., Wahr, J., Alexander, P., Box, J.E., Wouters, B. (2013) Evidence and analysis of 2012 Greenland records from spaceborne observations, a regional climate model and reanalysis data. *Cryosphere* 7, 615-630.
- Treguer, P., Nelson, D.M., Vanbennekom, A.J., Demaster, D.J., Leynaert, A., Queguiner, B. (1995) The silica balance in the world ocean a reestimate. *Science* 268, 375-379.
- Wadham, J., De'ath, R., Monteiro, F.M., Tranter, M., Ridgwell, A., Raiswell, R., Tulaczyk, S. (2013)

  The potential role of the Antarctic Ice Sheet in global biogeochemical cycles. *Earth and Environmental Science Transactions of the Royal Society of Edinburgh* 104, 55-67.
- WEHRMANN, L.M., FORMOLO, M.J., OWENS, J.D., RAISWELL, R., FERDELMAN, T.G., RIEDINGER, N., LYONS, T.W. (2014) Iron and manganese speciation and cycling in glacially influenced high-latitude fjord sediments (West Spitsbergen, Svalbard): Evidence for a benthic recycling-transport mechanism. *Geochimica et Cosmochimica Acta* 141, 628-655.
- WYNN, P.M., HODSON, A.J., HEATON, T.H.E., CHENERY, S.R. (2007) Nitrate production beneath a High Arctic Glacier, Svalbard. *Chemical Geology* 244, 88-102.
- ZHANG, J.Z., HUANG, X.L. (2011) Effect of temperature and salinity on phosphate sorption on marine sediments. Environmental Science & Technology 45, 6831-6837.

